Aging of di-isopropyl-phosphorylated human butyrylcholinesterase

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INTRODUCTION

Organophosphates are potent inhibitors of cholinesterases [acetylcholinesterase (AChE; EC 3.1.1.7) and butyrylcholinesterase (BuChE; EC 3.1.1.8)]; most of these compounds are used as insecticides, pesticides or drugs [1]. The most toxic organophosphates are chemical warfare agents [2]. The progressive inhibition of cholinesterases by organophosphates is due to phosphorylation (phosphorylation or phosphorylation) of their active site serine Ser-198 (200) in human (Hu) BuChE [3]. Phosphorylated cholinesterases react very slowly with water; they can be reactivated only by strong nucleophilic compounds such as oximes, which are used as antidotes against organophosphorus poisoning [1,4,5]. Organophosphoryl adducts can undergo dealkylation of an alkoxy group, which converts phosphorylated cholinesterases into non-reactivatable species, described as ‘aged’. Aging is a first-order process [6] that has been found to depend on the nature of the enzyme (BuChEs tend to age more rapidly than AChEs), the nature and stereochemistry of the organophosphate and the physicochemical conditions of the medium [7–9].

The generally accepted mechanism for the dealkylation reaction proposed by Michel et al. [10] involves a transient carbocation. This can result from protonation of the alkoxy oxygen [11], leading to scission of the C–O bond and formation of the carbonium ion and of a negatively charged monophosphoryl ester of serine, i.e. the aged enzyme [12–14]. This reaction can be depicted as in Scheme 1, where (a) is the free enzyme, (b) the organophosphate, (c) the covalent phosphoryl adduct that can be reactivated by oximes, (d) the developing carbonium intermediate, (e) the dealkylation products (R-OH, R-derived alkenes) resulting from the release of the carbonium ion into the medium (propan-2-ol is the major product when the organophosphate is di-isopropyl fluorophosphate (DFP) [15]) and (f) the monophosphorylate ester of serine [16]. Recently, a ‘push–pull’ mechanism of carbocation formation, involving both the glutamic residue (Glu-197) adjacent to the active site serine and the catalytic histidine residue (His-440) was proposed [17,18]. According to this mechanism, (1) formation of the carbocation is promoted by the strong negative electrostatic field of Glu-199 and (2) protonated His-440 acts as an acid catalyst for the cleavage of the C–O bond. Molecular dynamics calculations indicate that the anionic monophosphorylate ester is stabilized by hydrogen bond interactions with water and oxyanion hole residues and electrostatic interaction with protonated His-440.

The importance of protonated catalytic histidine (protonated

Abbreviations used: AChE, acetylcholinesterase; BCHE, butyrylcholinesterase gene, BuChE, butyrylcholinesterase; DFP, di-isopropyl fluorophosphate; DIP, di-isopropylphosphoryl; Hu, human; 2-PAM, 2-pyridinealdoxime methiodide. Note that italized numbers refer to the amino acid residues in Torpedo AChE that correspond to BuChE or AChE residues.

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His-440) in the aging process of both AChE and BuChE has long been recognized from pH-dependence studies [9,11] and from the observation that there is no aging when histidine is labelled by diethyl pyrocarbonate [15,19]. Several other active site residues involved in the dealkylation of soman-phosphonylated AChE have been identified [20–22]. In AChE, formation of the carbonium ion seems to be promoted by the glutamic carboxylate next to the phosphorylated serine residue (Ser-199) [17,20,22]. Other residues in the active centre have also been found to have a role in aging, including hydrogen-bond network residues Glu-450 (443), Tyr-133 (130) and Glu-202 (199) as well as Phe-338 (331) and Trp-86 (84) in HuAChE [22]. In both HuAChE [22] and HuBuChE [23] the aspartic residue located at the rim of the active site cleft (Asp-72) was found to be involved in the control of the aging process.

The molecular basis for the resistance to reactivation of aged cholinesterases may be ascribed partly to the electrostatic repulsion of the negatively charged oxygen of the P–Ocholinesterases may be ascribed partly to the electrostatic repulsion of the negatively charged oxygen of the P–O group. Evidence for a role in aging, including hydrogen-bond network residues Glu-450 (443), Tyr-133 (130) and Glu-202 (199) as well as Phe-338 (331) and Trp-86 (84) in HuAChE [22]. In both HuAChE [22] and HuBuChE [23] the aspartic residue located at the rim of the active site cleft (Asp-72) was found to be involved in the control of the aging process.

The molecular basis for the resistance to reactivation of aged cholinesterases may be ascribed partly to the electrostatic repulsion of the negatively charged oxygen of the P–O group and the adjacent negatively charged Glu-199 [18]. It also originates in conformational changes, concomitant with the dealkylation reaction, at the enzyme active-site gorge. Several works showed that aging is associated with cholinesterase conformational change [14,24] and increase in conformational stability [25–28]. This might result from the formation of a salt bridge between catalytic protonated histidine and the negatively charged oxygen atom on the monophosphylate moiety. Evidence for a salt bridge in the stabilization of aged BuChE is the pHK of 8.5–9 as estimated from differential scanning calorimetry of native and soman-aged BuChE at different pH values [28], and P. Masson, J. Clément, S. Rey and C. Cléry, unpublished work).

To determine the amino acids involved in the aging process, we expressed selected mutants of HuBuChE and determined their aging rate constants after phosphorylation by DFP. Mutations of Glu-197, mutation of the residue expected to interact with an alkoxy chain (Trp-82) and mutation of Asp-70 located at the rim of the active site cleft were all found to slow the aging rate process. To understand the effects of mutations, a molecular modelling study of the phosphorylated enzyme and its mutants was performed. Such an approach has proved to be very powerful in probing the active site of AChE by molecular dynamics of its non-aged soman adducts [29]. The present study revealed that mutations of Glu-197 had no drastic effect on the local structure of the non-aged adduct, so that the role of Glu-197 was not to maintain structure. Instead these results underlined the crucial role of the negative charge on Glu-197 in the dealkylation process. The residue belonging to the choline-binding site (Trp-82) was found to interact, mainly through van der Waals contacts, with an isopropyl chain of DFP. In particular, molecular dynamics of phosphorylated wild-type BuChE showed a stacking of the aromatic ring of Trp-82 against the putative leaving isoproxy group. This stacking was not observed in the simulation of the phosphorylated D70G mutant adduct, since the Ω loop of BuChE underwent a conformational change resulting in a shift of Trp-82 relative to the isoproxy group. Thus our results show that aging of DFP-phosphorylated BuChE involves amino acid residues equivalent to those promoting aging of soman-phosphonylated AChE, Trp-82 and Glu-197; moreover our results demonstrate that the remote Asp-70 exerts an indirect control on the aging process.

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Materials and Methods

Chemicals

DFP was purchased from Sigma Chemical Co. (St. Louis, MO, U.S.A.) and 2-pyridinealdoxime methiodide (2-PAM) was from Ega-Chemie (Steinteim, Germany). Racemic soman (pinacolyl methylphosphonofluoridate) was obtained from the Centre d’Etudes du Bouchet (Vert-le-Petit, France). Other chemicals were of biochemical grade. A stock solution of 10 mM DFP in anhydrous methanol acidified by 0.01% acetic acid was stored at –30 °C in borosilicate tubes.

Mutagenesis and Expression of HuBuChE

Site-directed mutagenesis of the HuBuChE gene (BCHE) cDNA was performed with PCR and Pfu DNA polymerase as described [30]. Mutated residues were Asp-70 (72), Trp-82 (84), Glu-197 (199), Ala-328 (330) and Phe-329 (331). Single and double mutants were made. Mutations of residues Asp-70, Trp-82 and Glu-197 were expected to slow the aging process. Mutations of residues Ala-328 and Phe-329 were made to test the possibility that the leaving isopropyl chain interacted with these residues. Eleven mutants were studied: A70G, W82A, E197D, E197Q, E197G, A328G, F329A, F329C, F329S, A328G+ F329S and E197Q+F329S.

Recombinant wild-type and mutant BuChEs were expressed in stably transfected Chinese hamster ovary K1 cells as previously described [30].
Purification of recombinant BuChE mutants

Recombinant enzymes were partly purified by affinity chromatography on a propanamide-Sepharose column [30], dialysed to remove propanamide and concurrently concentrated in an Amicon stirred cell with a PM10 membrane and stored at +4 °C in sterile 20 mM phosphate buffer, pH 7.0, containing 1 mM EDTA.

Enzyme activity measurements

BuChE activity was assayed with the Ellman method [31] with butyrylthiocholine iodide (1 mM) as substrate in 0.1 M sodium phosphate, pH 7.0, at 25 °C. Units of activity are μmol of butyrylthiocholine hydrolysed/min.

Enzyme phosphorylation and measurement of aging rates

Wild-type and mutant enzymes in 50 mM sodium phosphate, pH 8.0 (2–2.5 ml of approx. 0.2 unit/ml, i.e. max. 9 pmol of active sites) were inhibited to 90–95% by 30 μl of 20–100 μM DFP in methanol at 25 °C. The final methanol concentration (less than 2%, v/v) had no significant effect on enzyme activity. After enzyme had been inhibited, aliquots were removed at various times after inhibition (from 5 min to 7 days) and immediately incubated with 2-PAM (1 or 10 mM) in the same buffer at 25 °C. The time courses of reactivation were followed by measuring the activity recovered as a function of time of incubation with 2-PAM. Reactivation curves reached a plateau after 24 h of incubation. Reactivated enzyme activity was therefore measured after 24 h of incubation with 2-PAM. It should be noted that the dilution factor of reactivated enzyme in the Ellman assay was 1:150; under these conditions, 2-PAM, which is a competitive inhibitor (e.g. Ki = 0.20 mM for wild-type and Ki = 3.10 mM for D70G) of BuChE catalysed hydrolysis of butyrylthiocholine [23], did not significantly affect Kcat and thus did not interfere with activity measurements. Each experiment was done at least six times. The aging rates of wild-type and W82A soman adducts were determined in a similar way.

The rate constant of aging (ki) was determined from [32]:

\[
\log(100(E_a - E_i)/(E_a - E)) = -k_i t
\]

where Ea is the activity of control enzyme in the presence of 2-PAM, Ei is the residual activity after inhibition by DFP, E is the activity of reactivated enzyme and t is the time after inhibition by DFP and before the addition of 2-PAM. Results were also expressed as half-lives of aging: ti = ln(2/k). The half-life of aging of the wild-type BuChE was 28 min reported by Hobbiger [40] for aging of human plasma BuChE at pH 7.8 and 37 °C.

Molecular modelling and dynamics

The three-dimensional model of wild-type HuBuChE used as the starting structure was the Harel model [33]. All calculations were performed on an SGI Power Indigo 2 workstation and an SGI Power Challenge 4xR8000 super computer. Cartesian coordinates for BuChE were kindly provided by Dr. Harel [33]. Hydrogen atoms were added in INSIGHT 9.50 (Biosym, San Diego, CA, U.S.A.). All Arg, Lys, Asp, Glu, N-terminal and C-terminal groups were ionized except Glu-441, which is located in a hydrophobic environment. The missing fragment 482–486 was added by using the standard Biosym library; optimization of its geometry was achieved in DISCOVER 2.9 (Biosym). Minimization was performed with 200 steepest-descent steps followed by 200 conjugate-gradient steps while constraining all heavy atoms of the structure to their original positions [constant of 100 kcal/mol (418 kJ/mol)]. For mutants, replacement of the wild-type residue was achieved before minimization and the mutated residue was not constrained during the calculation. In a second step, Ser-198 was phosphorylated by DFP and His-438 was protonated. Charges for the phosphorylated serine residue were derived from calculations ab initio with the program DMol 2.3.5 (Biosym). The phosphoryl oxygen of DFP was placed in the vicinity of the oxyanion hole (Gly-116, Gly-117 and Ala-199). A grid on isopropyl torsion angles was then performed in QUANTA 96 (Molecular Simulations, Cambridge, U.K.) to minimize the electronic and steric interactions of isopropyl chains with the binding pocket. Finally, minimization was achieved by 500 steepest-descent steps followed by 500 adopted basis Newton–Raphson steps constraining all protein heavy atoms outside a shell of 20 Å from the phosphorylated serine [constant force of 40 kcal/mol (167 kJ/mol) between 20 and 22 Å, and 80 kcal/mol (335 kJ/mol) above 22 Å] with the molecular mechanics program CHARMM 23.2 [34].

Stochastic boundary molecular dynamics (SBMD) runs [35] were performed on DFP-phosphorylated wild-type and D70G mutant BuChE with a parallel version of the software CHARMM 23.2 (Molecular Simulations) [34]. The all-hydrogen model was used plus water was represented by the TIP3P type [36]. For non-bond interactions, a cut-off of 1 Å was applied with a switching function between 8 and 10 Å. The dielectric constant was set to 1. The reaction region was defined by a 23 Å sphere centred on the Ca atom of residue Ile-69. A buffer region was made of a 2 Å layer surrounding the reaction region. The remaining atoms of the protein constituted the reservoir region. The Ca of residue Ile-69 was chosen to include all the Ω loop atoms in the reaction region during the simulations. All non-hydrogen atoms of the protein in the buffer region were harmonically constrained with a boundary force. We used the simplified Debye–Wall factor values of Nakagawa et al. [37] in our calculations. The force was weighted by a switching function that decreased from 0.5 to 0 when moving from the buffer/reservoir boundary to the reaction region [35]. Solvation of the active site was achieved by using a pre-equilibrated sphere (25 Å radius) centred on the Ca atom of residue Ile-69. Water molecules overlapping the adduct (cut-off of 2.8 Å) were deleted. Four successive rotations of this sphere were applied, to minimize the remaining holes in the covalent complex. For the solvent boundary forces, a radial distribution function centred on the oxygen atoms was applied to the oxygen atoms [38]. The frictional coefficients related to the stochastic forces were assigned as follows: 62 ps for the heavy atoms in the protein and 200 ps for water oxygens [35,38]. SHAKE [39] was used to constrain bonds with hydrogen atoms, allowing a 2 fs time step. The SBMD simulation was performed by equilibrating the system at 300 K for 20 ps for each enzyme. Production dynamics of 200 and 800 ps were performed for non-aged phosphorylated wild-type and D70G mutant enzymes respectively. Coordinates were saved every 0.5 ps in each dynamics simulation.

RESULTS AND DISCUSSION

Aging of di-isopropylphosphorylated butyrylcholinesterase

To slow the aging process, inhibition of BuChE by DFP was performed at pH 8.0 and 25 °C [9]. The half-life of aging of the wild-type enzyme was found to be 60 min (Table 1). This value is consistent with t = 28 min reported by Hobbiger [40] for aging of human plasma BuChE at pH 7.8 and 37 °C and t = 3 h in water at 24 °C [16] for horse BuChE.
chloride as substrate (P. Masson and M. Th. Froment, unpublished work).

3.5.1.5 Molecular dynamics simulation of this adduct to see whether a conformational change in the indole ring of Trp-82 could explain the biochemical data. During a 200 ps trajectory, the aromatic ring of Trp-82 was found to stack against the isopropyl chain R₂ making close van der Waals contacts with the methyl groups (Figure 1, bottom panel). The conformational transition of Trp-82 arose from the disruption of the H-bond between H₂6 of Trp-82 indole ring and the hydroxy oxygen of Tyr-440, allowing the Trp-82 side chain to rotate closer to R₁. Interestingly, the second isopropyl chain (R₃) at the bottom of the active site was found to stack against Trp-231, leading to the sandwich-like structure depicted in Figure 1 (bottom panel). We believe that the effect of Trp-82 on the aging process originates in the stacking of the indole ring on the leaving alkoxy chain. In this position Trp-82 could polarize the O-C₁ bond or, most probably, stabilize the developing cation through interactions with the π electrons of the indole ring. Such interactions might also take place in the case of the soman-BuChE adduct, its aging process being faster than that of the DFP-BuChE adduct (t₉₀ = 9 min instead of 60 min) owing to a stronger inductive effect of the more substituted pinacolino moiety. The 10-fold difference in half-life between soman-phosphonylated W82A-BuChE (t₉₀ = 800 min) and W86A-AChE (t₉₀ = 7000 min) might, in contrast, result from additional interactions in the ‘anionic’ site of AChE that do not exist in the active site of BuChE.

The structural transition of Trp-82 did not affect the interaction of the phosphoryl oxygen with the oxyanion hole, nor did it change the distances to the three amide hydrogens of Gly-116, Gly-117 and Gly-199, which remained in an accepted range for such hydrogen bonds (see the time series in Figure 2). As expected, the strong hydrogen bond formed between Hₑ₂ of His-438 and the adjacent alkoxy oxygen during the minimization (see Figure 1) was found to remain during the time course of the dynamics (Figure 2), supporting the mechanism described in Scheme 1.

Concerning the acidic residue Glu-197, one of the side chain carboxylic oxygens is in the vicinity of carbon C-1 of the DIP moiety (average distance 4.47 Å). In addition, the carboxy oxygen of Glu-197 was found to interact directly with residue Glu-441 through a hydrogen bond with the hydroxy hydrogen of this non-charged residue (Figure 3). Though it has been shown for HuAChE [21] that the corresponding two residues are bridged by a water molecule, we found a direct interaction. This direct interaction was very effective in maintaining Glu-197 close to the isopropoxy chain (R₂) to be dealkylated in the simulation of the DIP-BuChE adduct. Because the side chain of Glu-197 is located near the centre of the active site in a region free of overlapping protein atoms, mutation of Glu-197 to an aspartate, glutamine or glycine residue did not lead to drastic conformational changes in the minimized structures of the phosphorylated mutant enzymes. These observations confirm that, as with AChE, Glu-197 mediates the aging process through its negative charge by stabilizing the carbion ion [17,20]. Moreover, the fact that the E197D mutation caused a decrease to one-sixth in the rate of aging relative to that of wild-type BuChE underlines the role of the Glu-197 side chain, whose size allows the carboxylate to be suitably placed for stabilizing the carboxylation.

We compared the final structure obtained by molecular dynamics calculations with that of the minimized soman-HuAChE adduct [22]. No important changes were found regarding the position of the catalytic His-438 (His-447 in HuAChE) with respect to the neighbouring Phe-329 (Phe-338 in HuAChE) in both adducts. Because we were unable to observe a meaningful decrease in the half-life of aging of DIP-BuChE for the F329A mutant, stabilization of the imidazolium ring of His-438 through a charge-π interaction with Phe-329 seems ques-

Effect of mutations on aging of DIP BuChE

Mutations of residues located in different areas of the active site cleft were found to affect the rate constant of aging of DFP-phosphorylated BuChE. As seen in Table 1, the most significant alterations were observed for mutations of residue Glu-197. Aging was slowed up to 155 times in the E197G mutant. Glu-197 has been found to stabilize the developing carbocation in soman-phosphorylated AChE [20–22]. Mutation W82A decreased the rate of the aging process with DFP to 1/15, and decreased the rate of aging with soman to 1/60 (t₉₀ = 13.5 h with racemic soman instead of BuChE adduct, aging at less than 9 min for wild-type enzyme; ratio ≈ 90).

<table>
<thead>
<tr>
<th>Enzyme</th>
<th>k₋₀ (min⁻¹)</th>
<th>t₉₀ (min)</th>
<th>t₉₀ ratio, mutant to wild-type</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wild-type</td>
<td>1130 ± 30</td>
<td>60 ± 5</td>
<td>1</td>
</tr>
<tr>
<td>D70G</td>
<td>140 ± 30</td>
<td>490 ± 40</td>
<td>8.0 ± 1.3</td>
</tr>
<tr>
<td>W82A</td>
<td>85</td>
<td>~ 810†</td>
<td>13†</td>
</tr>
<tr>
<td>E197D</td>
<td>190 ± 50</td>
<td>360 ± 100</td>
<td>5.8 ± 4.5</td>
</tr>
<tr>
<td>E197Q</td>
<td>20 ± 4</td>
<td>360 ± 65</td>
<td>60 ± 40</td>
</tr>
<tr>
<td>E197G</td>
<td>7.3 ± 0.9</td>
<td>9500 ± 110</td>
<td>155 ± 30</td>
</tr>
<tr>
<td>A328G</td>
<td>1310 ± 140</td>
<td>53 ± 2</td>
<td>0.9 ± 0.1</td>
</tr>
<tr>
<td>F329C</td>
<td>570 ± 20</td>
<td>120 ± 2</td>
<td>2.0 ± 0.2</td>
</tr>
<tr>
<td>F329A</td>
<td>580 ± 50</td>
<td>120 ± 10</td>
<td>2.0 ± 0.3</td>
</tr>
<tr>
<td>F329S</td>
<td>460 ± 60</td>
<td>150 ± 11</td>
<td>2.5 ± 0.3</td>
</tr>
<tr>
<td>A328G + F329S</td>
<td>265 ± 10</td>
<td>260 ± 7.0</td>
<td>4.2 ± 0.4</td>
</tr>
<tr>
<td>E197D + F329S</td>
<td>21 ± 3</td>
<td>3300 ± 400</td>
<td>54 ± 10.7</td>
</tr>
</tbody>
</table>

* t₉₀ = 12 min at pH 6.8 and 25 °C by using the ph-stat method with butyrylcholine chloride as substrate (P. Masson and M. Th. Froment, unpublished work).
† t₉₀ = 13.5 h with racemic soman instead of BuChE adduct.
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Figure 1 Stereo views of the wild-type BuChE active site in which Ser-198 is covalently attached to DIP

Van der Waals surfaces for side chains of residues Trp-82, Trp-231 and the phosphorylated serine are represented. The adducts are non-aged, that is, both isopropyl groups of DIP are present. The top panel shows the energy-minimized structure. The phosphoryl oxygen-binding pocket (oxyanion hole) is 9 Å away from the Trp-82 ring, preventing any interaction between the isopropyl chain R₁ and Trp-82, even in the most favourable orientation. The bottom panel gives the final structure of the 200 ps molecular dynamics simulation of the adduct, showing stacking of R₁ against Trp-82.

tionable. It is difficult to reach a conclusion on this point because the leaving alkyl chains of soman (pinacolyl) and DFP (isopropyl) are different in size. Nevertheless, no specific interactions between the R₁ isopropyl chain of DFP and Phe-329 could be observed in the energy-minimized structure of the DIP-BuChE conjugate, although we observed that the methyl moiety bound to the asymmetric carbon of the pinacolyl moiety pointed towards Phe-338 in the soman-HuAChE adduct. Thus van der Waals contact between this methyl and Phe-338 in the soman-HuAChE adduct might contribute to the formation of the carbocation.

Role of Asp-70 at the rim of the active site cleft

A puzzling question is how Asp-70 affects the rate of aging. A tentative explanation might involve the conformational state of the Ω loop. This loop, located between Cys-65 (67) and Cys-92
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Figure 2 Time series for the distances between the phosphoryl oxygen of DIP and the nitrogen atoms of the oxyanion hole (Gly-116, Gly-117 and Ala-199), and between N$_\varepsilon$2 of His-438 and the alkoxy oxygen of R$_1$.

Figure 3 Close-up of the average structure of the 200 ps dynamics showing the relative positions of residues Glu-197, His-438, Glu-441 and the phosphorylated Ser-198.

C$_1$ is a carbon in the R$_1$ isopropoxy group of DIP. The carboxylic oxygen O$_{\varepsilon1}$ of Glu-197 interacts with C$_1$, and with the uncharged hydroxy oxygen O$_{\varepsilon2}$ of Glu-441.

(94), has been shown to be involved in the allosteric modulation of AChE by peripheral anionic site ligands [41]. To test whether such an allosteric modulation could occur in aging reaction, we performed an 800 ps molecular dynamics simulation of D70G mutant bound to DIP. Despite the length of this run, we were unable to observe large conformational changes in the loop. Residues belonging to the $\Omega$ loop showed similar root-mean-square (RMS) fluctuations in the dynamics of both the wild-type and mutant enzyme adducts (Figure 4). However, disruption of the H-bond between Asp-70 carboxylate and Tyr-332 was found to affect the motion of the 70–77 segment, which underwent a slight displacement during the first few picoseconds of the simulation. This displacement induced a conformational change in Trp-82, whose indole ring shifted from its original position, moving away from the R$_{ij}$ isopropyl group and thus preventing the indole–isopropyl interaction in the D70G mutant. This could explain the slower (one-eighth) rate of aging of the DIP-D70G enzyme. Figure 5 compares the average structures of the dynamics of wild-type and D70G mutant adducts in this region of the enzyme structure. As can be seen, the global motion of the loop in the D70G mutant cannot be analysed in terms of lid opening as it can for Candida rugosa lipase [42]. Indeed, it has recently been shown that mutations of different residues in the $\Omega$ loop of AChE barely affected the catalytic properties of the enzyme [43]. As a consequence, allosteric modulation of the enzyme activity by the $\Omega$ loop would implicate small conformational transitions.

Figure 4 Root-mean-square (RMS) fluctuations of the $\Omega$ loop residues (backbone atoms) for the simulation of the phosphorylated wild-type enzyme (○) and for the first 200 ps dynamics of the phosphorylated D70G mutant (●).

Figure 5 Average conformation of the $\Omega$ loop in the two dynamics simulations: phosphorylated wild-type enzyme (blue), and phosphorylated D70G mutant (yellow).

Note the shift in position of Trp-82.
rather than large distortions in this region of the enzyme structure [44].

Conclusion and prospects
The single mutants E197G and E197Q and the double mutant 
E197Q + F329S enzymes, which have very slow aging rate 
constants, could be of biotechnological interest as bioscavengers 
for decontamination and detoxification or as parts of biosensors 
for organophosphate detection. Other double mutants, e.g. 
D70G + E197G, in association with oximes could be of phar-
macological interest as pseudocatalytic scavengers against 
organophosphate poisoning. The behaviour of the D70G mutant 
enzyme also raises the question of the relevance of the D70G 
mutant enzyme will function as an endogeneous 
activatable plasma enzyme will function as an endogeneous 
activatable plasma enzyme will function as an endogeneous 
pseudocatalytic organophosphate scavenger.

We thank Dr. A. Shafferman for the coordinates of the modelled structure of soman-
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